

## Question 1

1(d)(i)	covalent	1
1(d)(ii)	$  \begin{array}{c}  \text{H} \quad \text{H} \\    \quad   \\  \text{H}-\text{C}-\text{C}-\text{H} \\    \quad   \\  \text{H} \quad \text{H}  \end{array}  $	1
1(d)(iii)	chlorine	1
1(d)(iv)	2 (O <sub>2</sub> ) (1) CO <sub>2</sub> (1)	2

## Question 2

2(e)	methane	1
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## Question 3

3(d)	carbon dioxide	1
	water	1

## Question 4

4(d)	carbon dioxide (1)	2
	water (1)	

## Question 5

5(b)	carbon dioxide <b>AND</b> water	1
5(c)(i)	photochemical	1
5(c)(ii)	to provide the activation energy	1
5(c)(iii)	substitution	1
5(c)(iv)	2	1

## Question 6

6(e)(i)	$\text{C}_{10}\text{H}_{22} \rightarrow \text{C}_3\text{H}_8 + \text{C}_3\text{H}_6 + \text{C}_4\text{H}_8$	2
	<b>M1</b> $\text{C}_3\text{H}_8 + \text{C}_3\text{H}_6$ (1)	
	<b>M2</b> rest of equation correct (1)	
6(e)(ii)	cracking	1

## Question 7

7(a)(i)	needs or uses ultra violet light	1
7(a)(ii)	<b>M1</b> displayed formula of 1-chlorobutane(1)	2
	<b>M2</b> displayed formula of 2-chlorobutane(1)	